

AMIJIOL, ISOAMIJIOL, AND 14-DEOXYAMIJIOL, THREE NEW DITERPENOIDS  
FROM THE BROWN SEAWEED *DICTYOTA LINEARIS*

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Three new diterpenoids have been isolated from the brown seaweed *Dictyota linearis*, and their structures determined by  $^1\text{H}$  and  $^{13}\text{C}$  NMR spectroscopy.

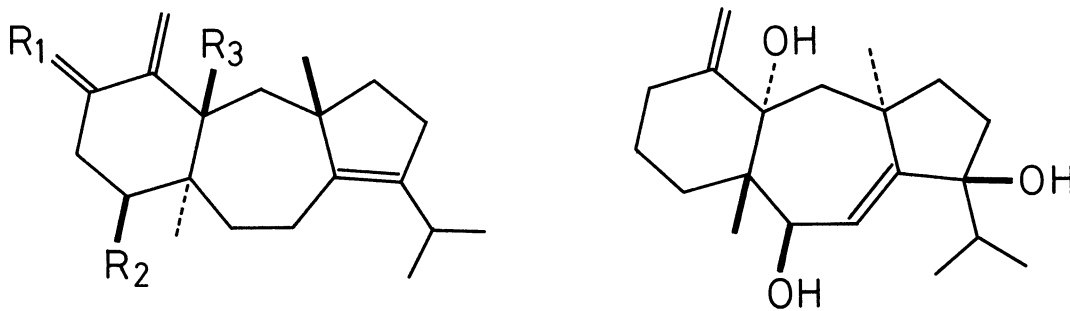
Recent investigations of the brown algae of the family Dictyotaceae have resulted in the isolation of the interesting secondary metabolites.<sup>1)</sup> During a search of the biologically active constituents of marine algae,<sup>2)</sup> we have examined the methanol extract of *Dictyota linearis* ("Itoamiji" in Japanese), which was found to have an antimicrobial activity against *Bacillus subtilis* and *Penicillium crustosum*. Careful silica gel column chromatography of the crude extract of fresh alga gave three new diterpenoids,<sup>3)</sup> amijiol (1), isoamijiol (2), and 14-deoxyamijiol (3) in 0.015, 0.026, and 0.002% yield respectively. We propose the linear tricyclic structures 1-3 for these compounds.<sup>4)</sup>

Amijiol (1),  $\text{C}_{20}\text{H}_{32}\text{O}_2$ , mp 180-181 °C, showed  $[\alpha]_{\text{D}} -126^\circ$  ( $\text{CHCl}_3$ ) and  $\nu_{\text{max}}$  ( $\text{CHCl}_3$ ) 3600, 3480, 3090, 1640, and 905  $\text{cm}^{-1}$ . The  $^{13}\text{C}$  NMR data for 1 assisted with off-resonance and selective proton-noise decoupling technique, as summarized in structure 1, showed the presence of four methyls, seven methylenes, one methine, one oxygen-bearing methine, together with three fully substituted carbon atoms, one tetrasubstituted double bond, and one exocyclic methylene.

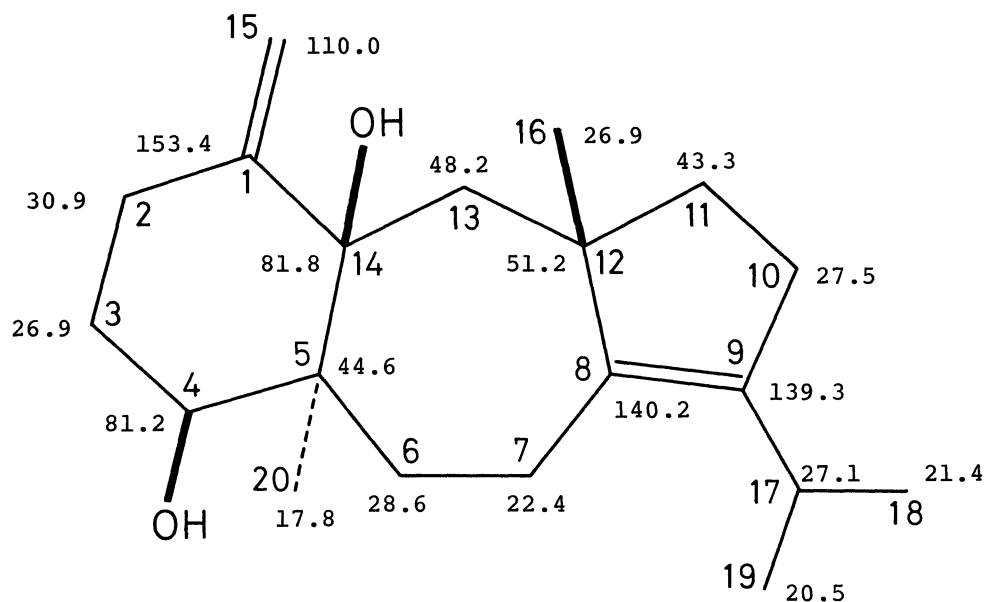
The pertinent  $^1\text{H}$  NMR data, completely assigned by an extensive decoupling study and NOE measurements, are shown in structure 1a; the gross structure with a linear 6-7-5 ring system was established by following observations. Four angular positions are occupied by two methyls, one hydroxyl group, and one double bond. The marked deshielding of 12-methyl would be reasonably interpreted through the consideration of the paramagnetic anisotropy by 14 $\beta$ -OH disposed in 1,3-diaxial relationship. In the  $^1\text{H}$  NMR spectrum of the acetate (4),  $\text{C}_{22}\text{H}_{34}\text{O}_3$ , mp 96-97 °C, the signal of 14 $\beta$ -OH proton appeared as a doublet ( $J=2.3$  Hz) at  $\delta$  3.64 by a long-range coupling with 13 $\alpha$ -H. This unusual W-coupling across C-O-H bonding should be caused by the hydrogen bonding to the carbonyl oxygen of 4 $\beta$ -OAc. The signal of 5-methyl exhibited a long-range coupling ( $J=\sim 0.1$  Hz) with 6 $\beta$ -H which was

considerably deshielded by two axial hydroxyl groups at C-14 and C-4. Observation of a 10% NOE on  $13\alpha$ -H upon irradiation of one of  $15$ -H signals revealed the presence of the contiguous  $C_1$ - $C_{14}$ - $C_{13}$  carbon atoms bearing substituents shown in structure 1. From these observations, we deduced the structure 1 for amijiol. This deduction is supported by consideration of the spectral data of the congeners, isoamijiol (2) and 14-deoxyamijiol (3), as follows.

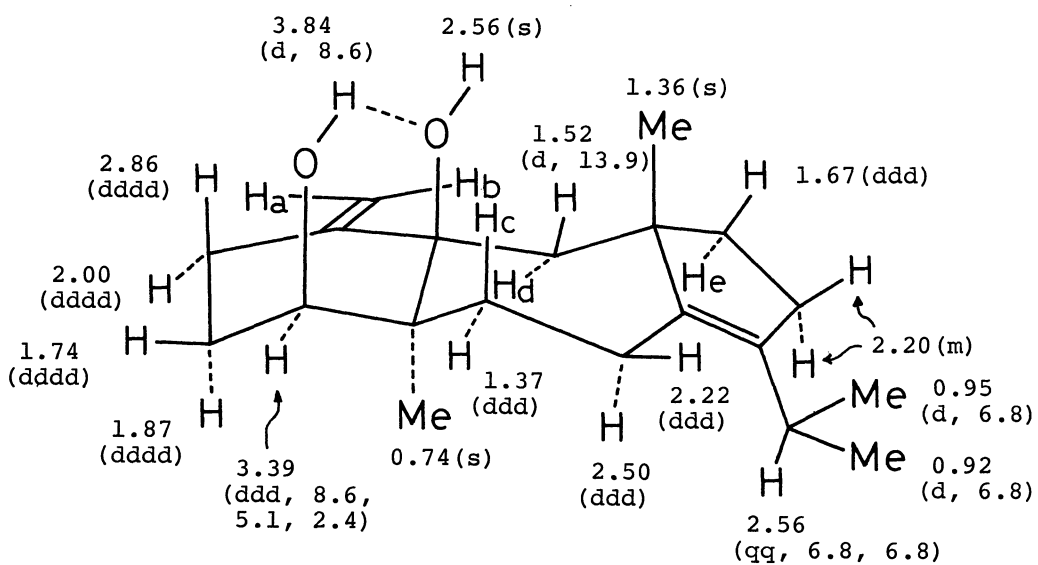
Isoamijiol (2), <sup>5</sup>  $C_{20}H_{32}O_2$ , mp 128-128.5 °C,  $[\alpha]_D -45^\circ$  ( $CHCl_3$ ), had the spectroscopic properties very similar to those of 1.  $\nu_{max}$  ( $CHCl_3$ ) 3600, 3480, 3090, 1640, and 885  $cm^{-1}$ ;  $^{13}C$  NMR ( $CDCl_3$ )  $\delta$  152.1 (C-1), 139.9 (C-8), 139.7 (C-9), 113.8 (C-15), 81.4 (C-14), 76.4 (C-2), 51.1 (C-12), 47.5 (C-13), 43.3 (C-11), 42.3 (C-5), 33.0 (C-4), 31.7 (C-6), 30.0 (C-3), 27.6 (C-10), 27.0 (C-17), 26.8 (C-16), 22.6 (C-7), 21.4 (C-19), 20.5 (C-18), and 17.8 (C-20);  $^1H$  NMR ( $CDCl_3$ )  $\delta$  5.09 and 5.02 (each br s,  $15$ -H<sub>2</sub>), 4.30 (ddd,  $J=3.4, 3.4,$  and  $0.5$  Hz, 2-H), 2.63 (qq,  $J=6.9$  and  $6.9$  Hz, 17-H), 1.34 (s, 16-H<sub>3</sub>), 0.95 and 0.92 (each d,  $J=6.9$  Hz, 18- and 19-H<sub>3</sub>), and 0.77 (s, 20-H<sub>3</sub>). The  $^{13}C$  and  $^1H$  NMR spectra of 2 and 1 are distinct from one another essentially only in the chemical shifts of the oxygen-bearing methine carbon and the methine proton attached to an oxygenated carbon atom respectively. This fact suggests that 2 has a structure close to that of 1, in which  $4\beta$ -OH in 1 is replaced by  $2\beta$ -OH. This was confirmed by the oxidation of 2 to yield an enone (5),  $C_{20}H_{30}O_2$ , mp 134-135 °C,  $\nu_{max}$  ( $CHCl_3$ ) 3600, 1695, and 1610  $cm^{-1}$ . Thus structure 2 is assigned to isoamijiol.



- (1)  $R_1 = H_2, R_2 = R_3 = OH$   
 (2)  $R_1 = \alpha\text{-H}, \beta\text{-OH}, R_2 = H, R_3 = OH$   
 (3)  $R_1 = H_2, R_2 = OH, R_3 = H$   
 (4)  $R_1 = H_2, R_2 = OAc, R_3 = OH$   
 (5)  $R_1 = O, R_2 = H, R_3 = OH$



(1) Amijiol;  $^{13}\text{C}$  NMR data (100 MHz,  $\delta$ /ppm) for  $\text{CDCl}_3$  solution.



(1a);  $^1\text{H}$  NMR data for 1; 400 MHz;  $\text{CDCl}_3$  solution;  $\delta$  values; multiplicity and J values (in Hz) in parentheses. Ha:  $\delta$  4.83 (dd, 1.5, 1.0), Hb:  $\delta$  4.88 (dd, 1.5, 1.5), Hc:  $\delta$  2.92 (ddd), Hd:  $\delta$  2.03 (d, 13.9), He:  $\delta$  1.57 (ddd).

A close structural similarity of 14-deoxyamijiol (3),<sup>5)</sup>  $C_{20}H_{32}O$ , mp 103-104 °C,  $[\alpha]_D -79^\circ$  ( $CHCl_3$ ), to 1 was indicated by the following spectral data analogous to those of 1.  $\nu_{max}$  ( $CHCl_3$ ) 3630, 3080, 1640, and 890  $cm^{-1}$ ;  $^{13}C$  NMR ( $CDCl_3$ )  $\delta$  152.8 (C-1), 140.8 (C-8), 139.7 (C-9), 107.6 (C-15), 79.4 (C-4), 50.0 (C-12), 43.6 (C-5), 42.3 (C-13), 41.3 (C-11), 40.3 (C-14), 36.8 (C-2), 31.8 (C-6), 31.5 (C-3), 27.5 (C-10), 26.9 (C-17), 24.2 (C-16), 22.4 (C-7), 21.2 (C-18), 21.0 (C-19), and 16.9 (C-20);  $^1H$  NMR ( $CDCl_3$ )  $\delta$  4.80 (ddd,  $J=1.6, 1.6,$  and  $1.3$  Hz, 15-Ha), 4.56 (ddd,  $J=1.6, 1.6,$  and  $1.6$  Hz, 15-Hb), 3.42 (dd,  $J=3.4$  and  $3.4$  Hz, 4-H), 2.67 (ddd,  $J=10.0, 6.0,$  and  $1.6$  Hz, 14-H), 2.59 (qq,  $J=7.0$  and  $7.0$  Hz, 17-H), 1.09 (s, 16- $H_3$ ), 0.93 and 0.92 (each d,  $J=7.0$  Hz, 18- and 19- $H_3$ ), and 0.65 (s, 20- $H_3$ ). These evidence reveal the replacement of the tertiary hydroxyl group at C-14 in 1 by a hydrogen atom. Therefore, 14-deoxyamijiol must be represented by the structure 3.

Structures 1 - 3 are closely related to that of dolatriol (6), a cytotoxic diterpenoid recently isolated from a mollusk *Dolabella auricularia*.<sup>6)</sup> The occurrence of amijiol and its congeners in a marine alga of the genus *Dictyota* may support the possibility that dolatriol is of dietary origin. Strange to say, three clavularane derivatives having a similar 6-7-5 ring system have been isolated from an octocoral *Clavularia inflata*<sup>7)</sup> which hardly feeds on such a seaweed.

We thank Dr. M. Ohno, Usa Marine Biological Institute, Kochi University, for identification of alga.

#### References and Notes

- 1) J. Finer, J. Clardy, W. Fenical, L. Minale, R. Riccio, J. Battaile, M. Kirkup, and R. E. Moore, *J. Org. Chem.*, **44**, 2044 (1979) and references cited therein.
- 2) For previous article see M. Ochi, H. Kotsuki, S. Inoue, M. Taniguchi, and T. Tokoroyama, *Chem. Lett.*, **1979**, 831.
- 3) These compounds did not show an evident antimicrobial activity against *B. subtilis* and *P. crustosum* at a concentration of 100  $\mu g/ml$ .
- 4) The structure of the congener, amijidictyol, was determined to have the same carbon skeleton by X-ray method; M. Ochi, M. Watanabe, M. Kido, Y. Ichikawa, I. Miura, and T. Tokoroyama, submitted for publication.
- 5) The detailed  $^1H$  NMR data for isoamijiol and 14-deoxyamijiol were consistent with the structures shown.
- 6) G. R. Pettit, R. H. Ode, C. L. Herald, R. B. Von Dreele, and C. Michel, *J. Amer. Chem. Soc.*, **98**, 4677 (1976).
- 7) J. C. Braekman, D. Daloz, R. Schubert, M. Albericci, B. Tursch, and R. Karlsson, *Tetrahedron*, **34**, 1551 (1978).

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